M.Sc.(Chemistry) CBCS Pattern Semester-III **PSCHT09 - Spectroscopy**

P. P Tim	ages : ie : Th	2 $GUG/W/23/113$ aree Hours* 6 3 4 5 *Max. Marks	331 : 80
1.	a)	Derive character table for C_3V point group on the basis of GOT? Identify point group for SF_4 molecule?	8
	b)	What is Great Orthogonality Theorem? Explain corollaries of the theorem?	8
		OR	
	c)	Identify point groups of following molecules?i) BCl_3 ii) NO_2 iii) BF_3 iv) C_2H_4	4
	d)	List elements of symmetry in: i) Pyridine ii) Cyclopropane iii) Ketene iv) 1, 3, 5-tribromobenzene	4
	e)	Show all rotation reflection operations are not considered as new operation.	4
	f)	Name the symmetry elements and symmetry operations in CH_4 molecule?	4
2.	a)	Explain the following: i) CI method for ion production. ii) Nitrogen rule.	8
	b)	Define "isomer shift" in Mossbauer Spectroscopy. Give its applications.	8
		OR	
	c)	Explain all possible fragmentation for in mass spectrometry.	4
	d)	How is Mossbauer used to determine the structure of $Fe(CO)_{12}$? Explain.	4
	e)	What is Metastable peak? Calculate position of metastable peak for toluene?	4
	f)	Discuss biological applications of Mossbauer Spectroscopy?	4

3.	a)	How does relative intensity of lines in rotational spectra depends upon the population of energy states? Discuss.	8
	b)	Discuss the following:i) Intensity of ESR lines.ii) Line width of signal.	8
		OR	
	c)	Calculate rotational constant 'B' for HCl molecule. The H-Cl bond length is 136 pm. The masses of 'H' and "Cl" can be taken as 1 and 35.5 respectively.	4
	d)	Explain applications of ESR spectroscopy.	4
	e)	Classify poly atomic molecules on the basis of moment of inertia about x, y, and z Axes?	4
	f)	Give an account of ESR spectroscopy of methyl radical and naphthalene.	4
4.	a)	Discuss the following: i) Morse potential energy function. ii) Isotopic effect in IR	8
	b)	Explain quantum mechanical theory of Raman effect?	8
		OR	
	c)	Explain classical theory for Raman Scattering?	4
	d)	Calculate force constant for HCl molecule if fundamental vibrational frequency. Is $8.667 \times 10^{13} \text{s}^{-1}$ (H = 1.008 amu and Cl = 35.5 amu)	4
	e)	Discuss how IR and Raman bands are complementary to each other.	4
	f)	Discuss Harmonic Oscillator.	4
5.	a)	List various symmetry elements in CH ₄ molecule.	2
	b)	Give example of molecule with $3\sigma v$ and $1\sigma h$.	2
	c)	Give principle of mass spectroscopy.	2
	d)	What is magnetic hyperfine interaction.	2
	e)	How is ESR used to study free radicals.	2
	f)	Give examples of symmetric top and asymmetric top molecules.	2
	g)	Calculate fundamental modes of vibration in NH ₃ molecule.	2
	h)	Explain Rayleigh Scattering.	2
