## M.Sc.- I (Chemistry) CBCS Pattern Semester-II

## **PSCCHT06 - Organic Chemistry**

P. Pages: 2 GUG/W/23/11229 Time: Three Hours Max. Marks: 80 Notes: 1. All questions are compulsory and carry equal marks. Explain the following reactions with mechanisms. 1. 8 a) Perkin reaction benzoin condensation ii) b) Explain-8 Michael addition reaction i) ii) Explain chemo selectivity with suitable example. OR c) Write a note on-4 Hydrolysis of amide Ammonolysis of ester What do you mean by electrophile and nucleophile? Give the mechanism and d) 4 stereochemistry of addition reaction involving electrophiles. e) Define regioselectivity. Give the mechanism of hydroboration-oxidation reaction. 4 Write the note on metal hydride reduction of unsaturated carbonyl compounds with suitable f) example. Explain the following rearrangement reactions with mechanisms. 2. 8 a) Hoffman rearrangement Pinacol-Pinacolone rearrangement ii) b) What do you mean by Free radicals. Discuss free radical substitution mechanism at an 8 aromatic and aliphatic substrate. OR Discuss various types of free radical reactions. c) d) Write a note oni) Wagner meerwin rearrangement Wolf rearrangement ii) Discuss the reactivity for aliphatic and aromatic substrates. 4 e) f) Give difference between Curtis and Schmidt rearrangements.

3.	a)	Discuss the Saytzeff's and Hoffman's rules in elimination reactions in details.			8
	b)	Explain the following terms- i) Autoxidation	ii)	Free radical rearrangement	8
			O	R	
	c)	Give the mechanism of E2 reaction.			4
	d)	Explain in brief- i) Reed reaction	ii)	Fenton's reagent	4
	e)	e) Discuss the effect of attacking base and leaving group on elimination reactions.			
	f) Why the reactions of NBS are highly regioselective? Discuss Sandmayerreaction				4
4.	a)	Discuss the following reactions in brief- i) Ugi reaction	ii)	Passereno reaction	8
	b)	Explains the green synthesis of- i) Paracetamol	i)	Ibuprofen	8
			O	R	
	c)	Explain the basic principles of green chemistry.			4
	d)	Write a short note on- i) Sonochemistry	ii)	Polymer supported reagents	4
e) What is nano chemistry? Explain nanotubes and nanorods.				nd nanorods.	4
	f)	How by the use of green chemistry the prevention and minimization of hazardous products take place.			4
5.	a)	Discuss the choice of solvents in green chemistry.			2
<ul> <li>b) Explain the need of green chemistry.</li> <li>c) Discuss the E1CB reaction.</li> <li>d) Explain Hunsdiecker reaction.</li> <li>e) Write the reactions of Tiffenev-Demjnov ring expansion.</li> </ul>					2
					2
					2
				expansion.	2
	f)	Write a note on neighbouring group assistance in free radical reactions.			
	g)	Explain catalytic hydrogenation of double bond.			2
h) Write Mannich reaction.					2

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