## M.Sc. - I (Environmental Science) CBCS Pattern Semester-I

## PSENVT01 / PSCENVT01-Paper-I : Environmental Chemistry

P. Pages: 1 Time: Thee Hours			irs * 6 1 8 5 *	GUG/W/23/11158 Max. Marks: 80	
	Note	otes: 1. All questions are compulsory and carry equal marks.  2. Illustrate the answers with suitable diagrams and examples.			
1.		Desc	ribe basic concept of colorimetry. Explain Lambert's law and Beer's	s law. <b>16</b>	
			OR		
	a)	Disc	uss principles of colloidal chemistry. Explain adsorption and absorpt	ion briefly. 8	
	b)	Expl	ain chemical potential and chemical equilibria.	8	
2.		Disc	uss alkalinity and acidity in detail.	16	
			OR		
	a)	Expl	ain buffering in water system.	8	
	b)	Desc	ribe oxidation and reduction processes.	8	
3.		Disc	uss physical characteristics of water.	16	
			OR		
	a)	Expl	ain hydrological cycle.	8	
	b)	Describe ionic product of water.			
4.		Expl	ain physical properties of soil.	16	
			OR		
	a)	Disc	uss soil profile with suitable diagram.	8	
	b)	Expl	ain soil formation processes.	8	
5.		a)	Explain molecular weight and equivalent weight.	4x4	
		b)	Comment on green chemistry for sustainable future.	=16	
		c)	Explain buffer solutions.		
		d)	Describe biofertilizers with suitable example.		
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