B.Sc.-III (CBCS Pattern) Semester - V USCCHT09 : Chemistry Paper-I (Organic Chemistry)

P. Pages: 2 Time: Three Hours			Max. Marks : 50	
1.	a)	Discuss principle of NMR spectroscopy. How many NMR peaks would you expect in i) Ethyl acetate ii) Acetone	5	
	b)	Write a note on.	5	
		i) Chemical shift		
		ii) Coupling constant (J)		
		OR		
	c)	Explain why TMS is used as reference compound in NMR spectroscopy.	21/2	
	d)	What do you mean by equivalent and non - equivalent proton in NMR spectroscopy.	21/2	
	e)	Explain Nuclear shielding & Deshielding.	21/2	
	f)	An organic compound having molecular formula C_3H_6O shows following NMR data i) 3H - triplet $\delta-1.5$ ii) 2H - Quartet $\delta-2.6$ iii) 1H - Singlet $\delta-9.6$ Deduce the structure of compound.	2½	
2.	a)	What do you mean by active methylene compound? Give mechanism of Claisen condensation?	5	
	b)	Discuss synthesis of Glycine and barbituric acids from diethyl malonate.	5	
		OR		
	c)	Discuss Acidity of α -hydrogen atom.	21/2	
	d)	Discuss synthesis of 4 - methyl Uracil from acetoacetic ester.	21/2	
	e)	Explain Keto - enol tautomerism.	21/2	
	f)	Give one method of preparation of diethyl malonate.	21/2	
3.	a)	Discuss any two types of polymerization reactions in detail.	5	
	b)	Discuss natural and synthetic Rubbers with example.	5	

	c)	What are polymers? Give its classification.	$2^{1/2}$
	d)	Discuss cross - linking polymerization reaction.	21/2
	e)	Explain Vulcanization of rubber.	21/2
	f)	Write a note on Conducting polymers.	21/2
4.	a)	Discuss twelve principles of green chemistry in detail.	5
	b)	Write a note on Green solvents and alternative methods in green chemistry.	5
		OR	
	c)	Explain catalysis in green chemistry.	21/2
	d)	Explain sustainable development using concept of green chemistry.	21/2
	e)	Discuss the term cradle to cradle in green chemistry.	21/2
	f)	Explain atom economy in green chemistry.	21/2
5.		Solve any ten.	1x10
		a) How many NMR peak obtain in Toluene.	
		b) Calculate δ value for a compound if it has \uparrow value 7.5?	
		c) Give any two solvents used in NMR.	
		d) Write two examples of active methylene compounds.	
		e) Structure of acetoacetic ester.	
		f) What is the nature of α -hydrogen in active methylene compounds.	
		g) Define Amphiphilic polymers.	
		h) Draw structure of Urea - formaldehyde resin.	
		i) Give examples of Biodegradable polymers.	
		j) Who is the father Green Chemistry?	
		k) What is fed stock in green chemistry?	
		l) Define solvent toxicity in green chemistry.	
