## M.Sc. (Chemistry) (CBCS Pattern) Sem-III PSCHT09 - Spectroscopy Paper-IX

P. P Tim	ages : ne : Th	2 ree Hours	* 2 0 5 1 *	GUG/W/22/11331 Max. Marks : 80	
1.	a)	Derive the character table for	or $C_2v$ point group using great Ort	hogonality theorem. 8	
	b)	Discuss the symmetry eleme	ents and symmetry operation in det	ail. 8	
			OR		
	c)	Discuss the classes of symmetry operation in $C_3v$ point group			
	d)	Explain the vibrational modes for $H_2O$ molecule using group theory.			
	e)	Derive the multiplication table for $NH_3$ .			
	f)	Explain reducible and irreducible representations.			
2.	a)	Discuss the different types of fission processed in mass spectrometry.			
	b)	Discuss the application of M i) Electronic structure	Iossbauer spectroscopy with respec ii) Surface studies	et to <b>8</b>	
			OR		
	c)	Discuss the fast atom bombardment method for ion production in mass spectrometry.		in mass spectrometry. 4	
	d)	The mass spectrum of ethylbenzene shows peaks at m/e values 106, 105, 91, 65. Explain predict the base peak.			
	e)	Discuss the basic principle of Mossbauer spectroscopy.			
	f)	Explain the following term i i) Isomer shift	n Mossbauer spectroscopy ii) Absorber	4	
3.	a)	Explain the following		8	
		i) Effect of substitution o	n transition frequencies.		
		ii) Stark effect.			
	b)	Explain the following in ES	R Spectroscopy.	8	
		i) Zero field splitting	ii) Kramer's degen	eracy	
			OR		

	c)	Explain the classification of molecules on the basis of moment of inertia.		4			
	d)	Discuss the diatomic molecule as a non-rigid	rotor.	4			
	e) Explain the ESR spectra of methyl and 1, 4-semibenzoquinone radical.						
	f)	f) How does the microwave spectra differ from IR spectra?					
4.	a)	Explain the Morse potential energy function of an anharmonic oscillator.					
	b)	Explain application of Raman spectroscopy for the study of active sites of metalloproteins.					
		OR					
	c)	<ul><li>Distinguish between following pair of compounds using IR spectroscopy.</li><li>i) Ethyl benzene and O-xylene</li><li>ii) Acetophenone and Benzyl alcohol</li></ul>					
	d)	Explain how Raman and IR are complementary to each other					
	e)	Discuss the rotational Raman spectra of a diatomic molecule.					
	f)	<ul><li>How would you distinguish between the following pairs on the basis of IR spectroscopy.</li><li>i) Fundamental vibration and overtones</li><li>ii) Inter and intramolecular hydrogen bonding</li></ul>					
5.	a)	Give the mathematic statement of great orthogonality theorem with the meanings of each symbol.					
	b)	Assign the point group. i) $H_2$ ii) iii) PCl <sub>5</sub> iv)	HCl $[Ni(CN), 1^{2-}]$	2			
	- )			2			
	c)	Write a short note on N-rule.					
	d)	Write a note on source in Mossbauer spectroscopy.					
	e)	Give the selection rule for microwave spectroscopy.					
	f)	Give the ESR lines in ESR spectrum of hydrogen free radical.					
	g)	How will you differentiate cis and trans isomer with the help of IR spectroscopy.					
	h)	What are the factors affecting the Raman peak intensities?					
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