M.Sc. S.Y. (Physics) (CBCS Pattern) Sem-III PSCPHYT11-1 - Core Elective E-1.1 - Materials Science-I

P. Pages : 1 Time : Three Ho		1 GUG/W/22/ ree Hours * 2 0 2 2 * Max. Mar	GUG/W/22/11298 Max. Marks : 80	
		Either:		
1.	a)	Explain stability and meta-stability in the context of Equilibrium and Kinetics.	8	
	b)	Discuss kinetics of thermally activated process.	8	
		OR		
	e)	What is Phase rule? Explain Binary phase diagrams with eutectic system.	8	
	f)	Explain Lever principle, super lattice and miscibility gap.	8	
		Either:		
2.	a)	Explain in detail Phase transformation in materials & Time scale for phase changes?	8	
	b)	Write a note on: (i) Peritectic reaction and (ii) Eutectoid and deutectic transformations.	8	
		OR		
	e)	Explain importance of dendritic structure in alloys.	8	
	f)	Explain transformation in materials on heating and cooling.	8	
		Either:		
3.	a)	What is meant by diffusion in solids? Explain Fick's law and their solutions.	8	
	b)	Explain diffusion of carbon in iron.	8	
		OR		
	e)	Explain Arrhenius theory of ionic conductivity.	8	
	f)	What is Kirkendall effect? Describe mechanism of diffusion.	8	
		Either:		
4.	a)	Explain in detail about construction and working of solar cell.	8	
	b)	Explain detail about Solid oxide fuel cells and phosphoric acid fuel cells.	8	
		OR		
	e)	What is Fuel cell? Explain advantages and disadvantages of fuel cell.	8	
	f)	Explain construction and working of proton exchange membrane fuel cell.	8	
5.		Attempt all of the following.		
		a) What are the Applications of zone refining?	4	
		b) Explain growth kinetics	4	
		c) Write a note on chemical sensors	4	
		d) Write short note biochemical fuel cell.	4	
