M.Sc. (Chemistry) (CBCS Pattern) Sem-I

PSCCHT02 - Paper- II : Organic Chemistry

P. Pages: 2 GUG/W/22/11184 Time: Three Hours Max. Marks: 80 All questions are compulsory and carry equal marks. Notes: 1. 1. 8 Explain the following terms. a) Phase transfer catalysis. ii) Crown ethers complexes. Explain conjugation and cross conjugation with example. 8 b) OR Explain any two synthetic applications of enamines. c) 4 d) Discuss inclusion compounds. What are imines? Give its application in organic synthesis. e) f) Explain Huckel's rule for benzoid compounds. 2. Explain optical activity of allenes and biphenyl compounds. a) b) Discuss singlet oxygen. Give its generation and reactions with organic substrates. OR Explain conformational analysis of monosubstituted cyclohexane. 4 c) d) Explain threo and erythron isomers. 4 Discuss the chemical reactions involving classical carbocations. e) Discuss the reactivity and formation of free radicals. f) **3.** Discuss neighbouring group participation by π bond with example. 8 a) b) Discuss the following terms. 8 Kinetic and thermodynamic control. i) ii) **Curtin-Hammett Principle** OR Explain Hammond's postulates. 4 c) d) Discuss the relation of Hammett equation with linear free energy. Explain intermolecular displacement reaction involving halogen atom. e) Discuss non classical carbocations in NGP. f)

4.	a)	Explain the following. i) Vilsmeier reaction ii) Pechman reaction.	8
	b)	Explain the effect of ambient substrate and ambient nucleophiles on vinylic carbon atoms.	8
		OR	
	c)	Discuss Sommelet-Hauser reaction.	4
	d)	Explain smiles reaction with its mechanism.	4
	e)	Explain effect of leaving group on SN_2 reaction.	4
	f)	Discuss Gattermann – Koch reaction.	4
5.	a)	Discuss aromaticity in tropylium cation.	2
	b)	What are graphene?	2
	c)	Define enantiotrotic and distereoprotic atoms.	2
	d)	Write the structure of nitrene.	2
	e)	Define isotope effect.	2
	f)	Explain in brief Taft equation.	2
	g)	Explain Diazo coupling reaction.	2
	h)	Write Von Richter reaction.	2
