e)	Explain the following.i) Stereochemical seriesii) Nephelauxetic effect
f)	Discuss the d – orbital splitting in tetragonally distorted complexes.
a)	Explain the SN^1 CB mechanism for the base hydrolysis with suitable example. Give the evidences in favour of SN^1 CB mechanism.
b)	Explain stepwise and overall formation constant of metal complexes. Discuss various factors affecting stability constant of metal complexes.
	OR
c)	Describe Jab's method for determining stability constant of complex
0)	Describe Job's method for determining stability constant of complex.
d)	Discuss the reaction without metal ligand bond breaking with suitable example.
d) e)	Discuss the reaction without metal ligand bond breaking with suitable example. Explain the potentiometric method for determination of formation constant.
 d) e) f) 	Describe Job's method for determining stability constant of complex. Discuss the reaction without metal ligand bond breaking with suitable example. Explain the potentiometric method for determination of formation constant. What is acid hydrolysis? Explain various stability constant of complex.
 d) e) f) a) 	 Describe Job's method for determining stability constant of complex. Discuss the reaction without metal ligand bond breaking with suitable example. Explain the potentiometric method for determination of formation constant. What is acid hydrolysis? Explain various stability constant of complex. i) What is mean by T (3c - 2e) bond and S(3c - 2e) in higher boranes? Explain with suitable example.

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- Explain the reduction of bond angle in $H_2O(104.5^\circ) > H_2S(92^\circ) > H_2Se(91^\circ)$
- d)
- Discuss Jahan Teller effect 2) OR Discuss Bent Rule and energetics of hybridisation. c)
- b) 1) What are bond pair, lone pair and geometry of following. XeO₄ SF₆ i) ii) iii) IF₇ XeO₂F₂ iv)

3.

P. Pages : 2

a)

1.

2.

Time : Three Hours

M.Sc. F.Y. (Chemistry) (CBCS Pattern) Sem-I **PSCCHT01 - Inorganic Chemistry Paper-I**

Explain crystal field splitting for square planar and trigonal bipyramidal complexes.

GUG/W/22/11183 Max. Marks: 80

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OR

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	c)	Explain classification of boranes.	4	
	d)	Discuss STYX number for higher boranes and their utilities.	4	
	e)	Describe detail structure and bonding in $B_{10}H_{14}$	4	
	f)	Give the preparation of carboranes.	4	
	a)	What is iso and heteropoly acids? Explain the Keggins theory with structure?	8	
	b)	What is metal cluster? How are they classified? Write brief account of bonding in metal cluster.	8	
OR				
	c)	Explain the stereochemistry aspect of Re metal cluster.	4	
	d)	Give two examples of tetranuclear complexes of acetate cluster.	4	
	e)	What is metal – metal bond? Explain its formation considering molecular orbital theory.	4	
	f)	Discuss the applications of poly acids.	4	
	a)	Give the name of following compounds	2	
		i) B_5H_{11} ii) B_2H_2		
		11) D 5118		
	b)	Write the limitation of CFT.	2	
	c)	Give the classification of carboranes.	2	
	d)	Draw structure of $\operatorname{Re}_2 \operatorname{Cl}_8^{2-}$ cluster.	2	
	e)	Write a short note on chelation effect.	2	
	f)	Explain the various spectrochemical rules for VSEPR theory.	2	
	g)	Give two examples of isopoly acids.	2	
	h)	Explain the effect of metal ion on stability of complex.	2	

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